



Cambridge IGCSE™

COMBINED SCIENCE

0653/51

Paper 5 Practical Test

May/June 2020

MARK SCHEME

Maximum Mark: 40

Published

Students did not sit exam papers in the June 2020 series due to the Covid-19 global pandemic.

This mark scheme is published to support teachers and students and should be read together with the question paper. It shows the requirements of the exam. The answer column of the mark scheme shows the proposed basis on which Examiners would award marks for this exam. Where appropriate, this column also provides the most likely acceptable alternative responses expected from students. Examiners usually review the mark scheme after they have seen student responses and update the mark scheme if appropriate. In the June series, Examiners were unable to consider the acceptability of alternative responses, as there were no student responses to consider.

Mark schemes should usually be read together with the Principal Examiner Report for Teachers. However, because students did not sit exam papers, there is no Principal Examiner Report for Teachers for the June 2020 series.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the June 2020 series for most Cambridge IGCSE™ and Cambridge International A & AS Level components, and some Cambridge O Level components.

This document consists of **7** printed pages.

Generic Marking Principles

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptors for a question. Each question paper and mark scheme will also comply with these marking principles.

GENERIC MARKING PRINCIPLE 1:

Marks must be awarded in line with:

- the specific content of the mark scheme or the generic level descriptors for the question
- the specific skills defined in the mark scheme or in the generic level descriptors for the question
- the standard of response required by a candidate as exemplified by the standardisation scripts.

GENERIC MARKING PRINCIPLE 2:

Marks awarded are always **whole marks** (not half marks, or other fractions).

GENERIC MARKING PRINCIPLE 3:

Marks must be awarded **positively**:

- marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate
- marks are awarded when candidates clearly demonstrate what they know and can do
- marks are not deducted for errors
- marks are not deducted for omissions
- answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.

GENERIC MARKING PRINCIPLE 4:

Rules must be applied consistently e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

GENERIC MARKING PRINCIPLE 5:

Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).

GENERIC MARKING PRINCIPLE 6:

Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

Science-Specific Marking Principles

- 1 Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
- 2 The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
- 3 Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
- 4 The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.

5 'List rule' guidance

For questions that require *n* responses (e.g. State **two** reasons ...):

- The response should be read as continuous prose, even when numbered answer spaces are provided
- Any response marked *ignore* in the mark scheme should not count towards *n*
- Incorrect responses should not be awarded credit but will still count towards *n*
- Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response
- Non-contradictory responses after the first *n* responses may be ignored even if they include incorrect science.

6 Calculation specific guidance

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form, (e.g. $a \times 10^n$) in which the convention of restricting the value of the coefficient (*a*) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

7 Guidance for chemical equations

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

Please read the MARK SCHEME STANDARDISATION – EXPLANATORY DOCUMENT before authoring your Mark Scheme, and ensure that you are using the correct template for your subject/qualification/component. Please delete this paragraph after reading.

Question	Answer	Marks
1(a)	green / blue AND matching pH ;	1
1(b)(i)	(surface area) cube B = 600 AND cube C = 150 ;	1
1(b)(ii)	(volume) cube B = 1000 AND cube C = 125 ;	1
1(c)(i)	time (to change colour to red) AND / s ;	1
1(c)(ii)	all values recorded ; values consistent with candidate's heading ; cube C less time than cube A ;	3
1(c)(iii)	increase in surface area to volume ratio takes less time to change / ORA ;	1
1(c)(iv)	cubes not cut exactly / difficult to judge end point / AVP ;	1
1(c)(v)	cutting AND use knife, on solid surface / away from body ; OR acid / indicator AND use, gloves / goggles ;	1
1(d)(i)	suitable scale chosen ; all points plotted correctly ;	2
1(d)(ii)	suitable curve of best fit ;	1

Question	Answer	Marks
2(a)	(hydrochloric acid) red AND (sodium hydroxide) yellow AND (A) orange ;	1

Question	Answer	Marks
2(b)(i)	colour and volume for experiment 1 ; all colours and volumes ; 2 and 3 within 0.4 cm^3 ;	3
2(b)(ii)	(volumetric) pipette ;	1
2(b)(iii)	none that are red / the closest ones / not the first AND get more accurate after first / those within 0.2 or 0.4 cm^3 ;	1
2(b)(iv)	correct average calculated using only chosen values ;	1
2(b)(v)	average calculated in (b)(iv) $\times 3$;	1
2(c)(i)	(more concentrated solution) hydrochloric acid AND less used than sodium hydroxide ;	1
2(c)(ii)	expect four times more concentrated with some type of working out ;	1
2(d)(i)	white ; powder / solid / residue ;	2
2(d)(ii)	white ppt. ;	1

Question	Answer	Marks
3(a)	line drawn along normal out of straight edge labelled X ;	1
3(b)(i)	angle of refraction recorded for 10° ;	1
3(b)(ii)	angles recorded for 20° and 30° ; angle recorded for 40° AND no refraction for 50° ; angle of refraction increasing as angle of incidence increases ;	3
3(c)	(estimate of critical angle =) answer greater than 40° and less than or equal to 50° ;	1
3(d)	measure the angle of refraction for angles of incidence in between 40° and 50° / carefully move the light source around the curve back and forth to find the smallest angle of incidence where no light is refracted and measure this angle of incidence ;	1

Question	Answer	Marks
4	<p><i>one marking point from each section and any <u>three</u> others:</i></p> <p>method and apparatus fill tank with water to a specified depth (< 20 cm) ; use of stopwatch / suitable timing device ; repeat and take average <u>for each depth</u> ;</p> <p>variables keep method of producing waves the same ; same water source / same environment ; minimum of four values of depth of water given ; values of depth have a range of at least 15 cm AND do not exceed 20 cm maximum depth ;</p> <p>results time for the waves to travel length of tank AND depth of water ;</p> <p>conclusion calculation of speed of waves from the time measurements ; use results to plot graph of depth against speed ; look for pattern in the change of speed as depth increases ;</p>	7